



Assessment of Toxic Risks from Warehouse Fires

Graham Atkinson, Principal Scientist

HSE Science and Research Centre, Harpur Hill, Buxton SK17 9JN

Brian Briggs, HM Specialist Inspector

HSE Chemicals, Explosives & Microbiological Hazards Division,
Redgrave Court, Bootle, L20 7HS

Why is the assessment of fume risks from chemical warehouses important ?

1. It is a legal responsibility under COMAH
2. It can motivate and guide improvements ... fire precautions, segregation, stock reorganisation ...
3. It can support a “Let Burn” policy where this is appropriate.

History of regulation

- 1984 Control of Industrial Major Accident Hazards (**CIMAH**) Regulations implemented the European Communities “Seveso” Directive
- 1992 Extension of CIMAH Regulations to cover sites that simply stored toxic materials i.e. chemical warehouses.
- 1999 Current COMAH Regulations also apply to chemical warehouses



Sandoz Fire - 1st November 1986

COMAH Regulations

SCHEDULE 4

PART 1: PURPOSE OF SAFETY REPORTS

....**2.** demonstrating that major accident hazards have been **identified** and that the **necessary measures** have been taken to prevent such accidents and to limit their consequences for persons and the environment;

PART 2: MINIMUM INFORMATION

....**4.** **assessment of the extent and severity** of the consequences of identified major accidents;

Could there be toxic risks from fumes (in principle)?

Simple (worst case) assumptions

- Dangerous dose for paraquat $7 \times 10^{-5} \text{ kg/m}^3$
- Mass of active in warehouse 500,000 kg
- Maximum volume of “fatal” cloud $7.1 \times 10^9 \text{ m}^3$

If such a cloud were 1 kilometre wide and 100 m high it would stretch for 71 kilometres.

Excessively conservative. More realistic modelling is required.

Important issues in warehouse fire risk assessment:

- Frequency
- Timing of fire growth/Structural response
- Toxic release rate
- Buoyancy of fumes
- Dispersion
- Toxic effects
- Representing a large complex inventory

Almost all of these are complex problems and there is very little technical guidance available.

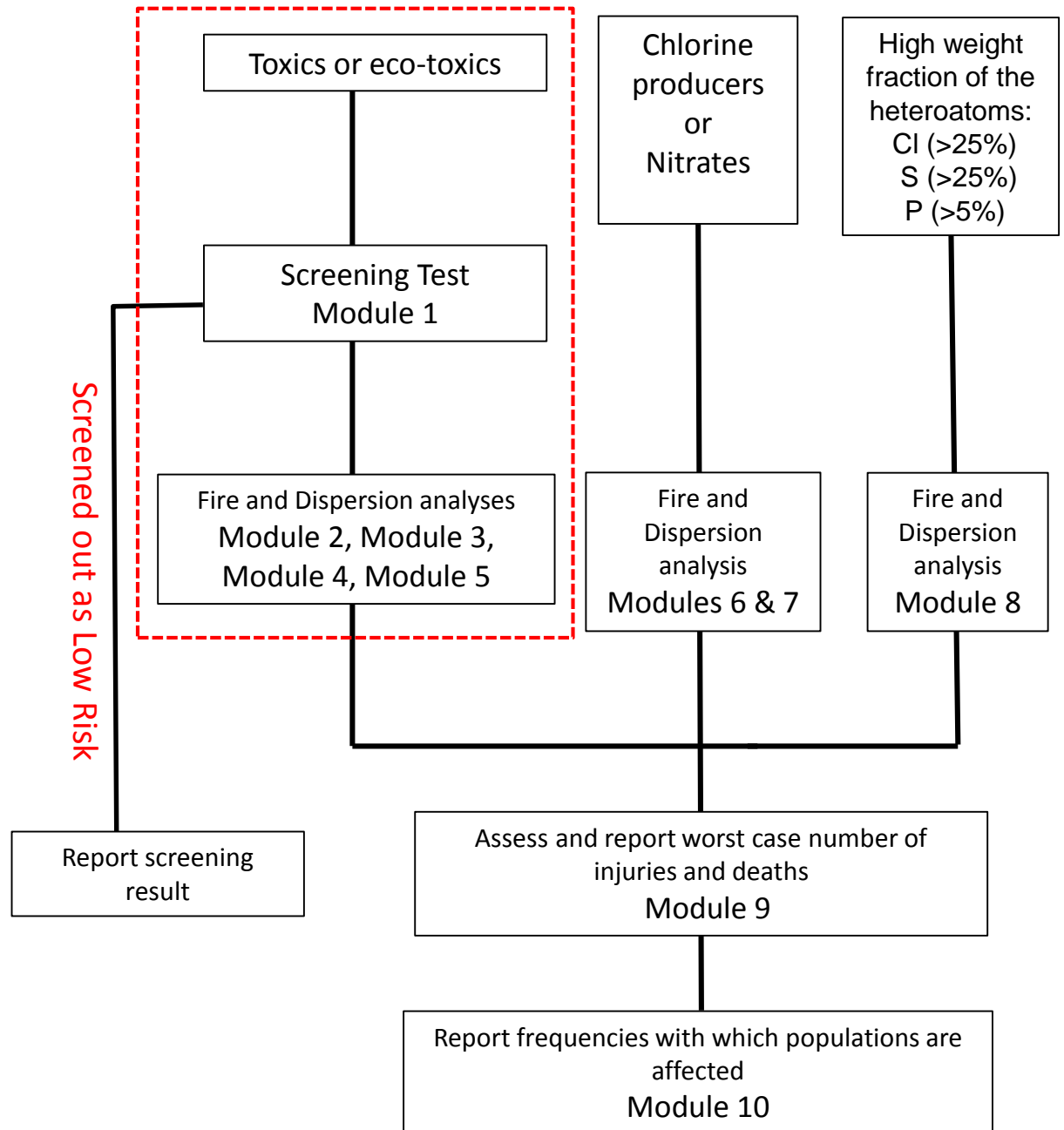


Ventilation controlled fire no external flaming

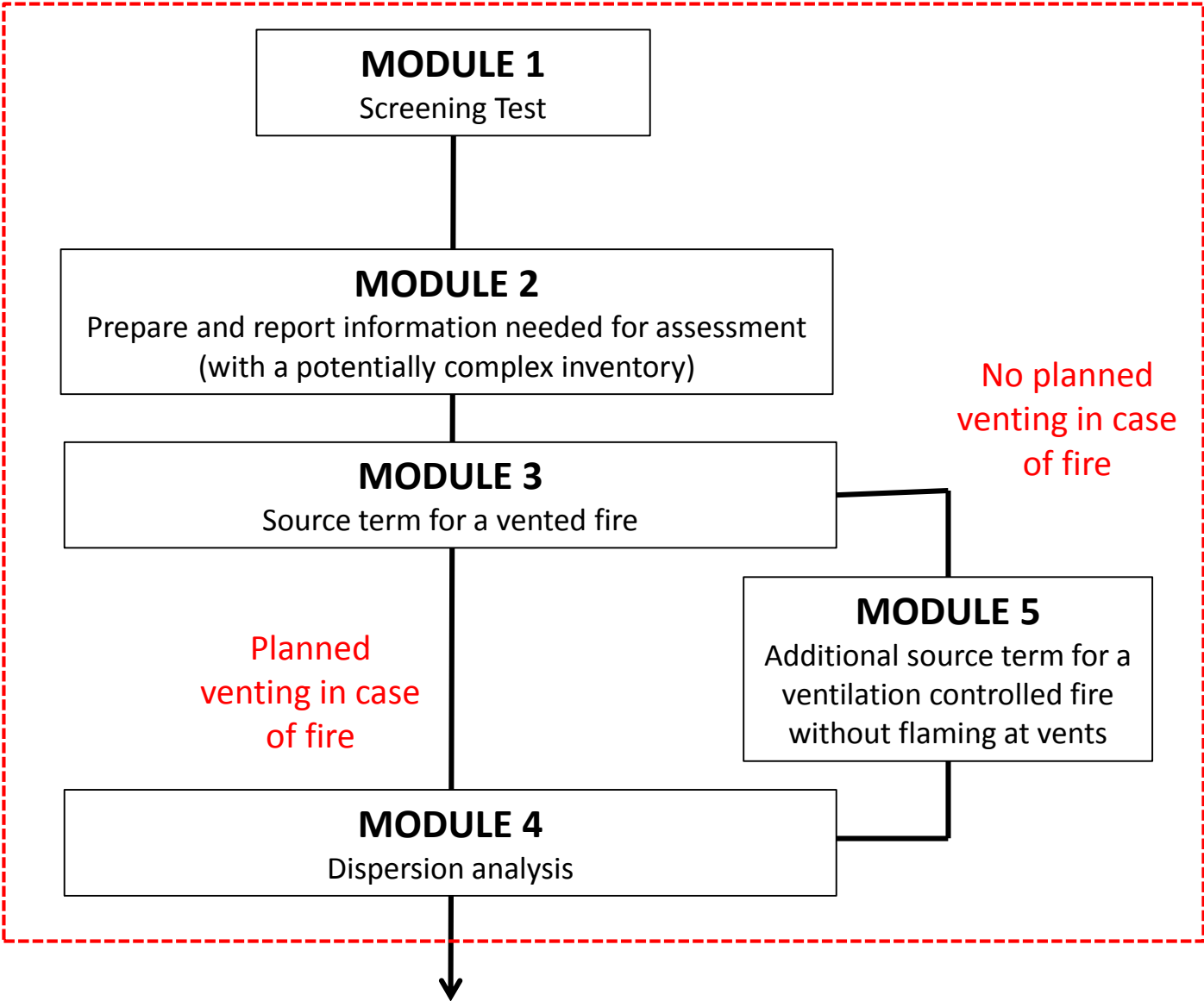
Well ventilated external flames



Structure of the proposed method



Main modules for toxics



Dealing with complex toxic inventories

You can use:

1. HSE “A” and “n” values

<http://www.hse.gov.uk/chemicals/haztox.htm> .

(Good for a small number of common single component chemicals)

Or

2. Classification data under CLP

<https://echa.europa.eu/guidance-documents/guidance-on-clp>

(More practical if there are a large number of mixtures e.g. agrochemicals.)

LC₅₀ (4 hour) levels to be assumed in assessment

Products classified acutely toxic Category 4 (Hazards statement H332, H302)	1 mg/l
Products classified acutely toxic Category 3 (Hazards statement H331, H301)	0.5 mg/l
Products classified acutely toxic Category 2 (Hazards statement H330/2, H300)	0.05 mg/l
Products classified acutely toxic Category 1 (Hazards statement H330/1, H300)	LC ₅₀ (4 hour)

How are LC₅₀ values used to define levels of harm?

An exposure time of 30 minutes is assumed

Definition of SLOD (exposure causing **death**)

$$SLODconc \left(\frac{mg}{l} \right) = LC_{50}(30min)$$

or

$$SLODconc \left(\frac{mg}{l} \right) = LC_{50}(4\ hour) \times 8$$

Definition of SLOT (exposure causing **injury**)

$$SLOTconc \left(\frac{mg}{l} \right) = \frac{LC_{50}(30min)}{4}$$

or

$$SLOTconc \left(\frac{mg}{l} \right) = \frac{LC_{50}(4\ hour)}{4} \times 8$$

Reducing a complex inventory to an equivalent dispersible mass

Example agrochemicals warehouse Product	Total Holding (kg)	Dispersible mass (10% dispersal) (kg)	Adjustment factor	Class 4 Equivalent dispersible mass
H330/H300-Cat1 substances	None	None	-	0
H330/H300-Cat2 substances	15,200	1,520	20	30,400
H331/H301-Cat3 Substances	47,500	4,750 kg	2	9,500
H332/302-Cat4 Substances	87,100	8,710 kg	1	8,710
Total				48,610

Assessment of the generation of heat and dispersal of toxic materials

There is always a complex risk balance between generation of heat, release of toxic materials and plume lift off.

A high fire load increases plume lift-off and reduces effects at ground level...

But it tends to increase the efficiency with which toxic materials are dispersed.

How is this balance reflected in the method?

Heat generation

The fire consumes all of the combustible contents of the warehouse in a period of 3 hours.

Plume Dispersion

The ground level concentrations fall off with increasing heat generation

Dispersal of toxic materials

“Toxic materials that are not combustible and are stored in non-combustible packaging will not contribute to fumes if they are segregated horizontally from combustible materials” ... (definitions) ... “and isolated from burning pools caused by leakage from other goods in a large fire.”

For low (non-zero) fire loads, low dispersal fractions also usually apply.

On balance

Overall, a very low fire load warehouse is likely to be associated with a lower risk – if it is sensibly organised.

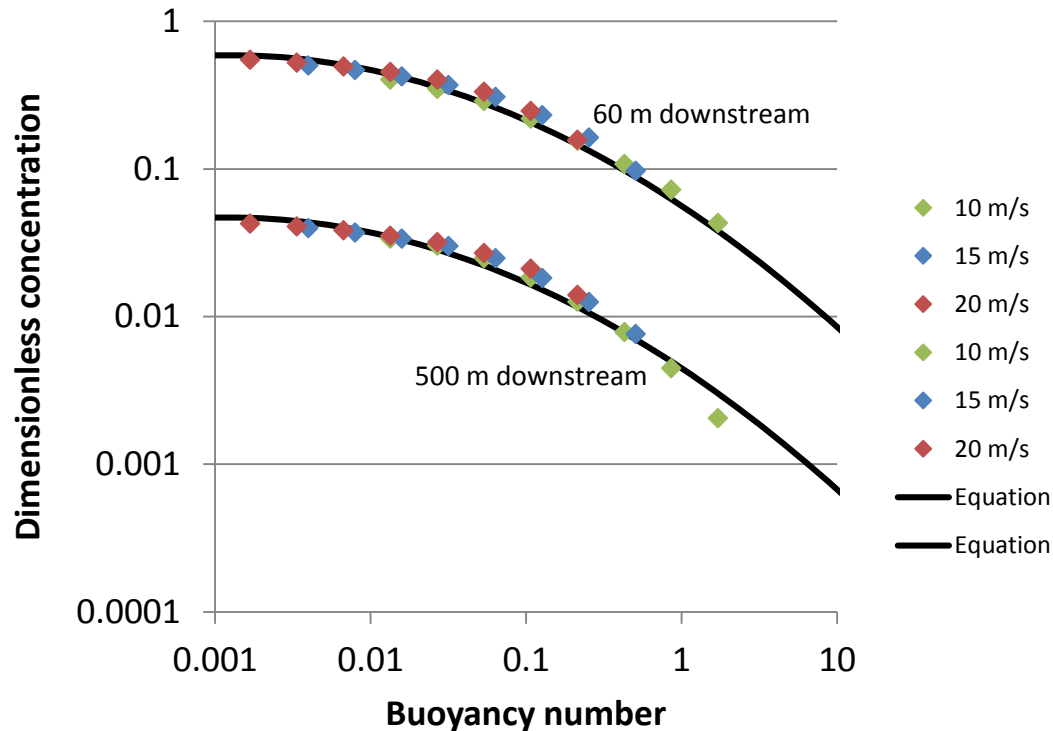
Dispersion is controlled by the buoyancy number

$$B = \frac{\text{Heat} \cdot 26.7 \cdot Q(\text{MW})}{\text{Wind}^3 \cdot \text{Size} \cdot W}$$

A single equation is provided to calculate concentrations downwind

$$C = 0.17 \cdot \frac{\dot{M}_{\text{toxic}}}{U \cdot W^2} \cdot \left(\frac{W}{R}\right)^{1.2} \cdot 10^{-(0.7 \log_{10} B + 0.12 (\log_{10} B)^2)}$$

No separate dispersion model needed



Plume seeding (release) fractions

Factors increasing risk

High level storage
Small package sizes
Fine powders
Pressurisable
containers
High thermal stability
Restricted ventilation¹

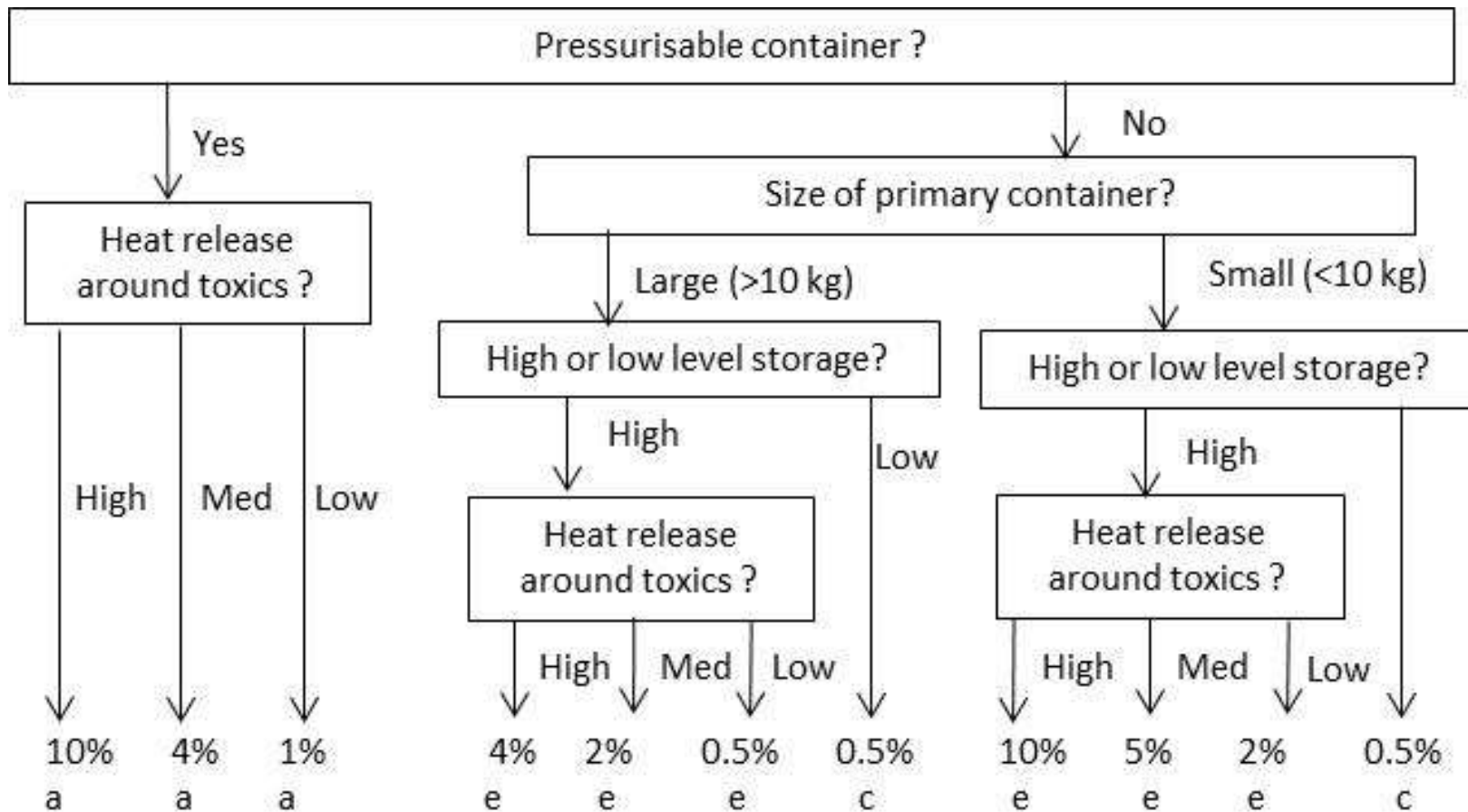
Factors reducing risk

Ground level storage
Large package sizes
Non dispersible materials
Non-pressurisable containers
Decomposition before boiling
Flammable toxic material
Recirculation (in the warehouse)²

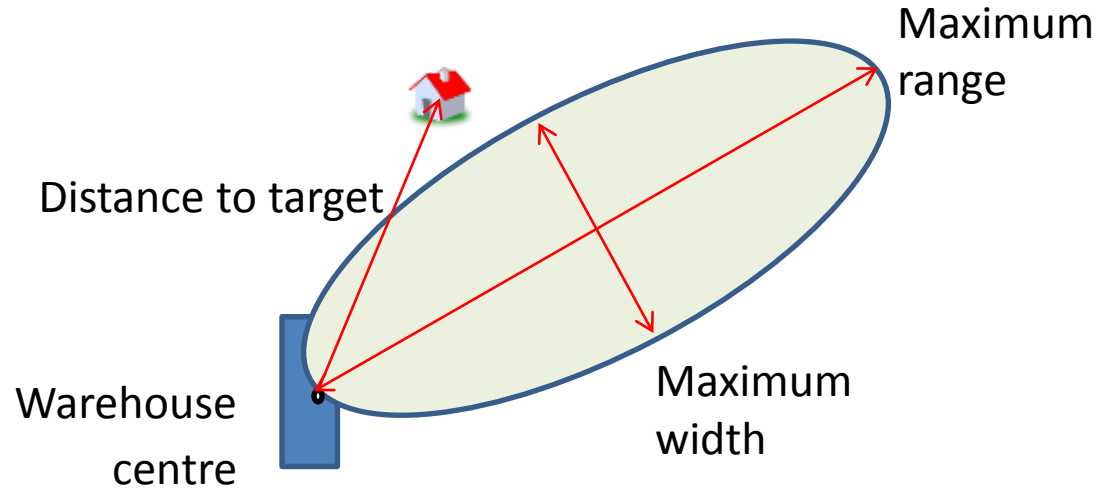
¹For volatile (and combustible) toxic materials – e.g. pesticides

²For non-volatile , incombustible toxic materials – e.g. heavy metal powders

Liquid products



Calculating plume shape and probability of exposure



$$\text{Maximum width (m)} = 0.75 (\text{Maximum range} - m)^{0.75}$$

$$R = \text{Distance to target} / \text{Maximum plume range}$$

$$\text{Probability of exposure } p = 1/\pi \cdot \arctan 0.46 R \cdot (R-R^2)^{1/2}$$

Ratio R (Target distance/Maximum plume range)	0.1	0.2	0.3	0.4	0.5	0.6	0.7	0.8	0.9
Probability of exposure (uniform wind rose)	0.300	0.236	0.194	0.163	0.137	0.114	0.093	0.071	0.048

Wind speed probabilities (inland sites)

Wind Speed (m/s)	Probability
5	0.5
10	0.1
15	0.03

Example wind direction data (London) – Prevailing wind is SW

Wind direction	Probability (relative to a uniform wind rose)
W	1.95
SSW	1.58
WSW	1.31
WNW	0.951
NW	0.804
NNW	0.634
N	0.756
NNE	0.756
NE	0.707
ENE	0.853
E	0.682
ESE	0.609
SE	0.560
SSE	0.707
SSW	1.26
S	1.85

Incident history – harm to people

Warehouse fires are amongst the most common and destructive events to affect chemical storage sites **worldwide**; sometimes large quantities of toxic and high toxic materials are destroyed. However, **the authors know of no offsite fatalities linked to such events** and it must be the case that this would only occur in unusual circumstances i.e. a particularly toxic inventory, unfavourable weather, high population density etc.

The purpose of warehouse fire assessment is to identify such high risk cases and generally to direct efforts to reduce risk. Such mitigation measures include: providing planned ventilation in the event of fire; separating toxic materials from combustible goods where possible and storing dispersible toxics at ground level.

Incident history – harm to the environment

There are many examples of chemical warehouses fires that have caused major environmental damage through contaminated firewater run-off. One use of fire plume toxicity assessment is to support “let burn” decisions in planning for and dealing with large fires.

The future

This method has been more fully documented (with numerous examples) than is possible at Hazards 29. The method may be of interest to warehouse trade associations and other bodies. The authors would be willing to discuss its use should such bodies wish to adopt or develop the method.