

Priority Topic Area: Environmental Protection and Improvement

Research Background

As greenhouse gas (GHG) emissions continue to rise rapidly (**Figure 1**), addressing climate change remains an urgent global challenge. While CO₂ is the most abundant GHG, **methane** and **nitrous oxide** (N₂O) have far greater **global warming potentials** (28 and 265 that of CO₂ respectively)¹. N₂O emissions stem from nitrogen-based fertilisers in **agriculture**, as well as key **industrial processes** such as adipic and nitric acid production. Alongside supporting decarbonisation efforts, this study highlights how **small-scale reactor testing** can enhance technical understanding of N₂O abatement through heterogeneous catalysis processes.

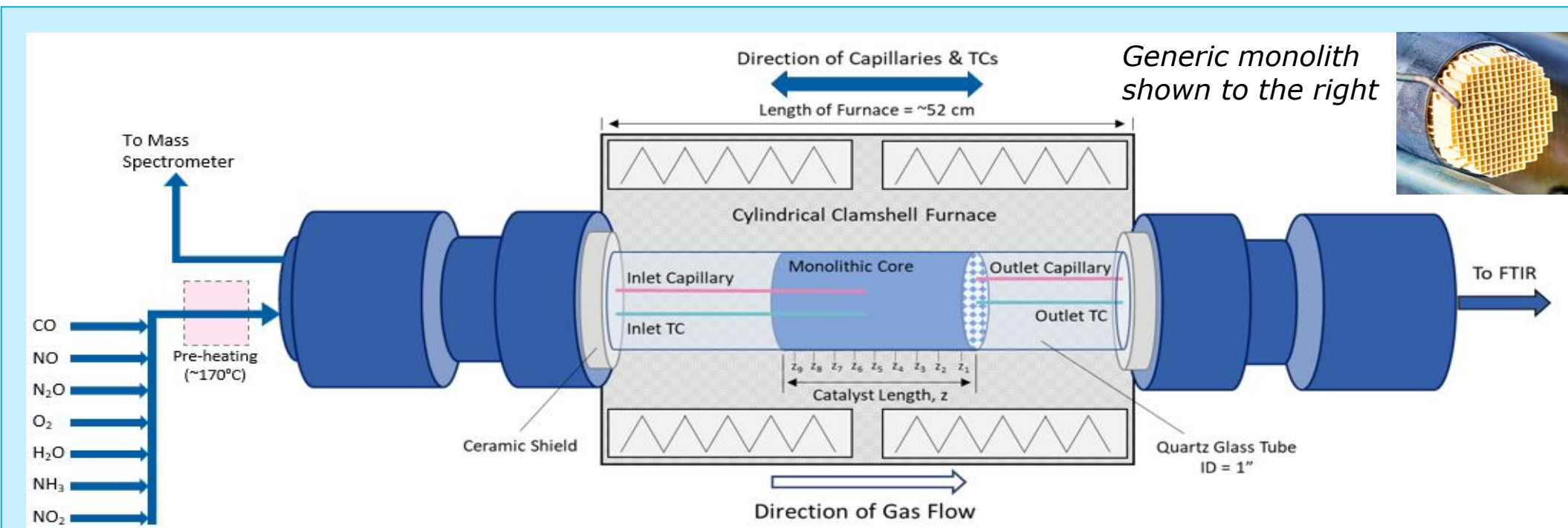


Figure 2: A schematic displaying the SPACI-MS experimental rig [figure adapted from reference 2]

Experimental Findings

- Simulated **Catalyst Activity Testing (SCAT)** revealed effect of different process variables, including inlet temperature and gas concentrations
- Results showed the **advantageous effect** of **increasing NO** concentration on N₂O decomposition.
- **Figure 3** demonstrates how **SPACI-MS** supported this study, confirming the **beneficial effect of NO**, and showing the spatial resolution of N₂O conversion, which ranged from **~14% to >90%** by **5cm** along the core
- Corresponding temperature profiles showed exothermic increases from **42°C to 198°C**, highlighting regions of **catalytic activity** and **thermal loads** on the catalyst.
- **Figure 5** contains a possible mechanism indicating how NO works to **reduce Fe-active sites** to **accelerate N₂O decomposition**.
- **Figure 4** used SCAT testing results to build a **predictive model** demonstrating the effect of varying inlet gases on N₂O conversion.
- Testing done within the context of **ammonia engines**, under **Innovate UK-funded project REGGAE**, validated previous trends observed.
- Early results suggest **NH₃** has **parabolic effect** on N₂O conversion, as seen in **Figure 4**.

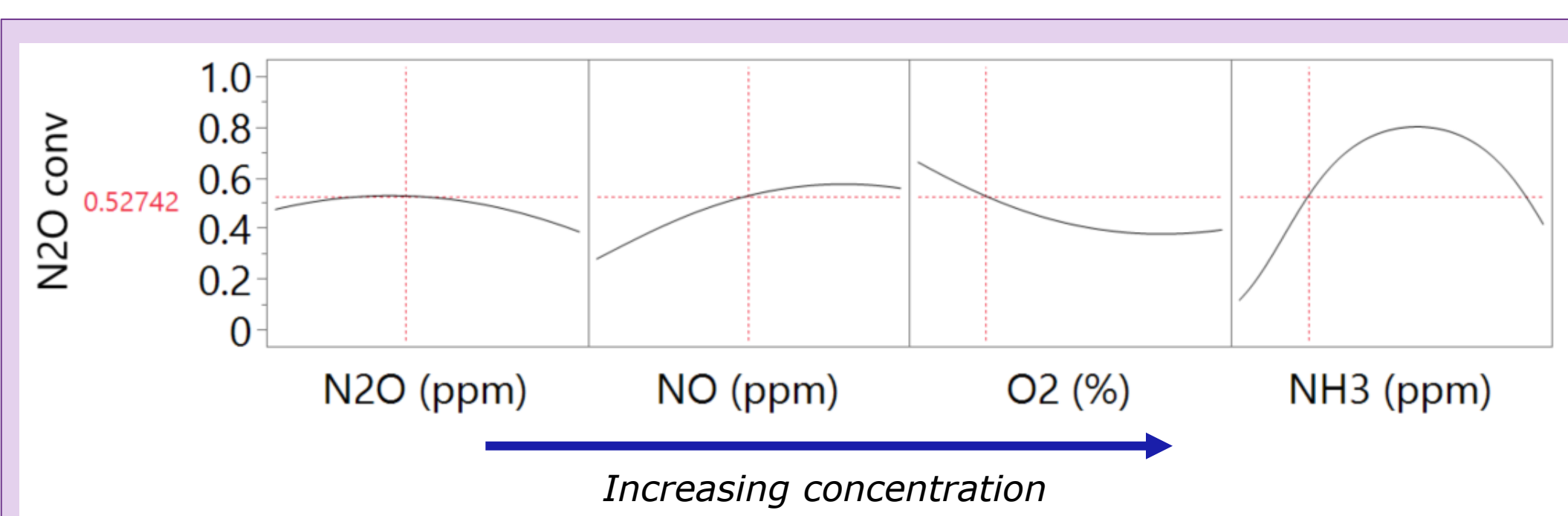


Figure 4: The interactive DoE profiler which shows the main trends observed: N₂O concentration has no significant effect bar a slight negative trend; NO has a promoting effect and O₂ is demoting. There appears to be an optimum concentration of ammonia.

The Importance of Understanding N₂O Decomposition

N₂O has a global warming potential 265¹ times that of CO₂ and is currently the most harmful gas to the ozone layer⁴. It is therefore paramount to abate as much N₂O as possible to contribute towards meeting globally agreed upon climate change targets. Research like this shows how current catalytic technology can help companies improve on sustainability targets as well as phasing in newer, greener fuels.

Acknowledgements

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References

1. Hannah Ritchie, Pablo Rosado, and Max Roser (2020) - "Greenhouse gas emissions" Published online at OurWorldinData.org. Retrieved from: 'https://archive.ourworldindata.org/20260223-070107/greenhousegasemissions.html' [Online Resource] (archived on February 23, 2026).
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3. Alves, L., Holz, L. I. V., Fernandes, C., Ribeirinha, P., Mendes, D., Fagg, D. P., & Mendes, A. (2022). A comprehensive review of NOx and N2O mitigation from industrial streams. *Renewable and Sustainable Energy Reviews*, 155, 111916. https://doi.org/10.1016/j.rser.2021.111916
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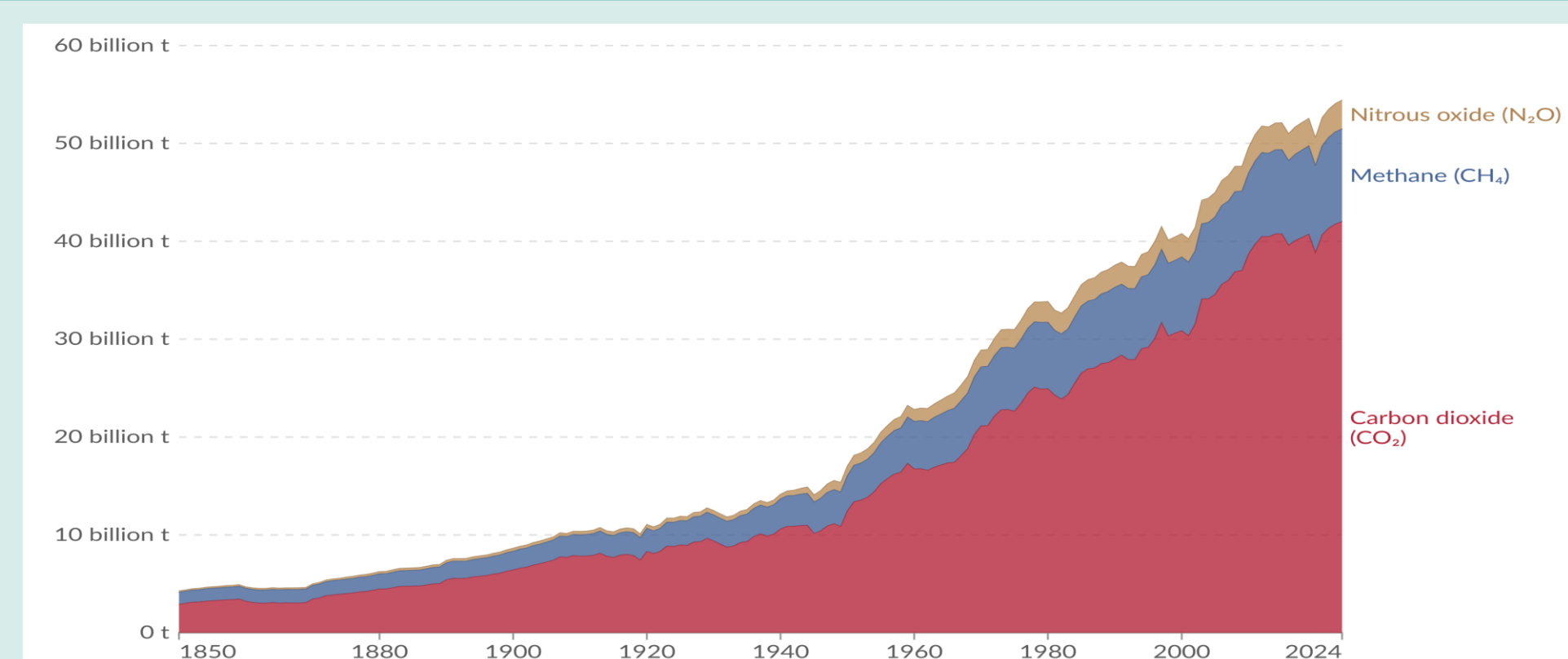


Figure 1: A graph displaying a breakdown in the increase in GHG emissions from 1850-2024 (Ritchie, Roser and Rosado, 2024)

Testing Methodology

- Extensive literature studies and JM's knowledge of **N₂O abatement** for automotive applications with **PGM catalysts** and **metallic zeolites** determined economic viability of an **iron-based zeolite** for N₂O abatement.
- After conducting a rigorous **Design of Experiment (DoE)**, key process variables were rationalised, tested and optimised, whilst minimising the number of experiments.
- Experiments were carried out using **monolithic reactors**, which simulate gas exhaust conditions.
- Using **SPACI-MS** (**S**patially **R**esolved **C**apillary **I**nlet **M**ass **S**pectrometry, **Figure 2**) with an FTIR analyser, evolutions of reactions, and **concentration** and **temperature** profiles throughout catalysts were uncovered.

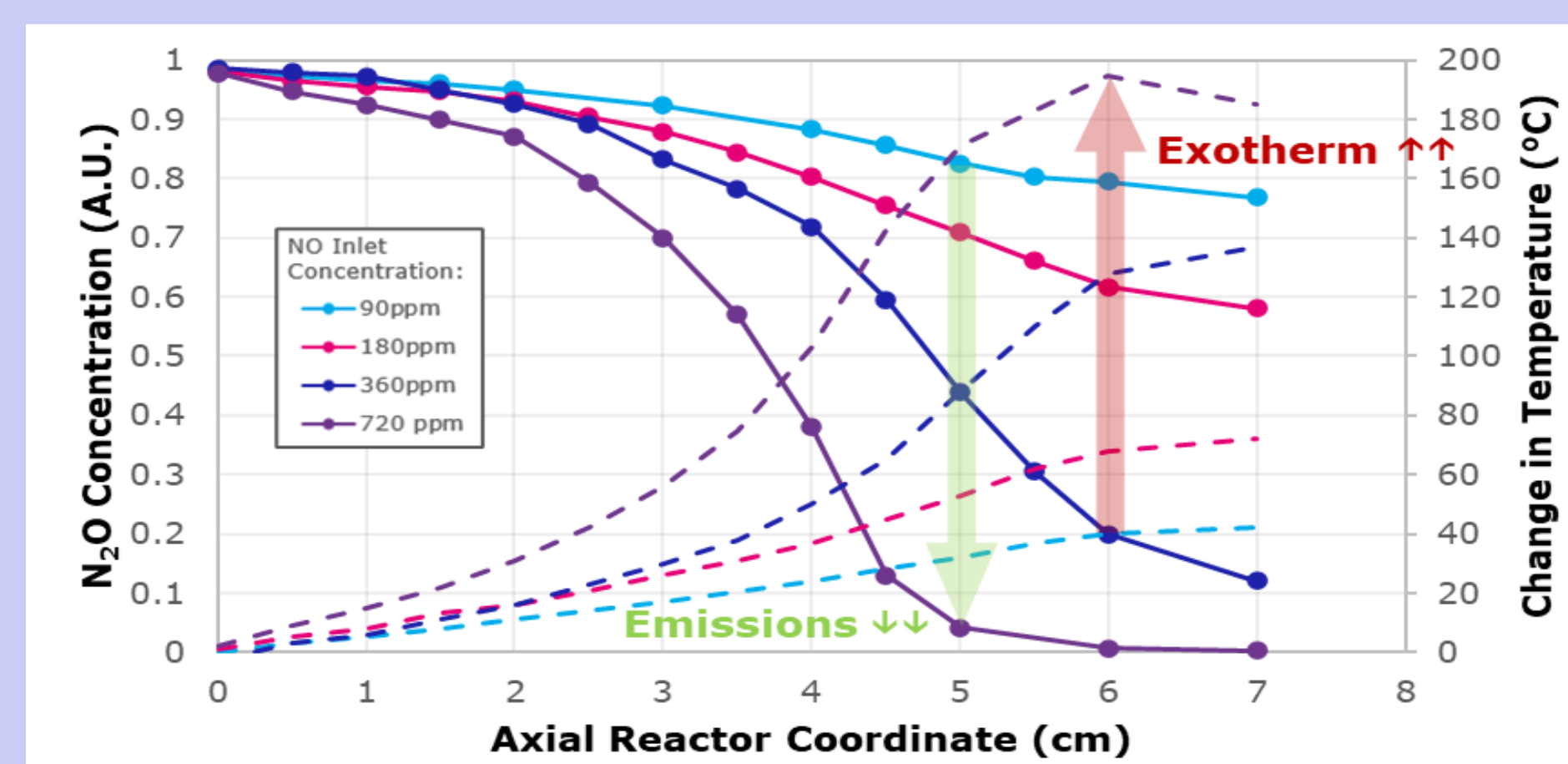


Figure 3: A graph displaying the change in N₂O concentration along the axial length of the catalyst, according to varying NO inlet concentrations. The corresponding temperature profiles are indicated with dashed lines

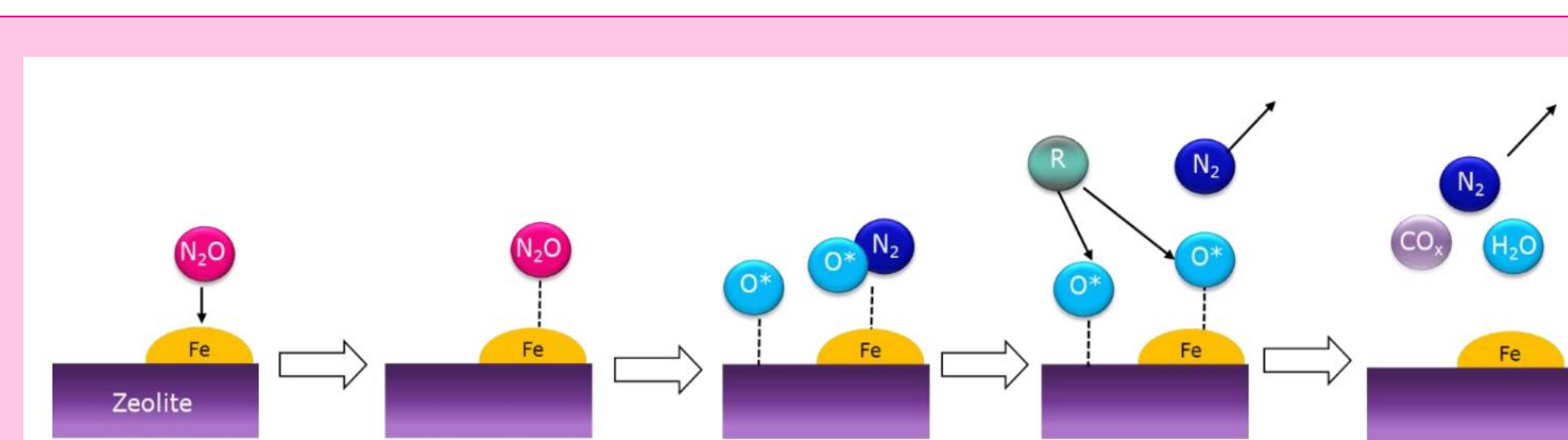


Figure 5: Schematic showing the mechanism for N₂O decomposition whereby R is the reductant (either NO, NH₃, H₂ or a hydrocarbon) and that various products can be formed [figure adapted from reference 3].

Next Steps & Future Development

Testing like this shows promising potential for extension from traditional autocatalysis applications to future ones, like data centres, wastewater treatment, or alternative internal combustion engine fuels for marine applications. There is further room for novel formulations and catalysts to be tested to accommodate more stringent emissions requirements.